This article was downloaded by:

On: 26 January 2011

Access details: Access Details: Free Access

Publisher Taylor & Francis

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-

41 Mortimer Street, London W1T 3JH, UK



## Nucleosides, Nucleotides and Nucleic Acids

Publication details, including instructions for authors and subscription information: http://www.informaworld.com/smpp/title~content=t713597286

# Synthesis of Peptide Nucleic Acid Monomers

L. Kovács<sup>a</sup>; Z. Timár<sup>a</sup>; B. Penke<sup>a</sup>

<sup>a</sup> Department of Medicinal Chemistry, A. Szent-Györgyi Medical University, Szeged, Hungary

To cite this Article Kovács, L. , Timár, Z. and Penke, B.(1999) 'Synthesis of Peptide Nucleic Acid Monomers', Nucleosides, Nucleotides and Nucleic Acids, 18: 4, 727-729

To link to this Article: DOI: 10.1080/15257779908041555 URL: http://dx.doi.org/10.1080/15257779908041555

### PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.informaworld.com/terms-and-conditions-of-access.pdf

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

### SYNTHESIS OF PEPTIDE NUCLEIC ACID MONOMERS

L. Kovács\*, Z. Timár, B. Penke

Department of Medicinal Chemistry, A. Szent-Györgyi Medical University, Dóm tér 8., H-6720 Szeged, Hungary. E-mail: kovacs@ovrisc.mdche.u-szeged.hu

**ABSTRACT**: The chemical synthesis of peptide nucleic acid (PNA) monomers is described using Fmoc (backbone), anisoyl (cytosine, adenine), 4-tert-butylbenzoyl (cytosine) and isobutyryl/diphenylcarbamoyl (guanine) protecting group combinations. For the guanine monomer the alkylation was realized both in a Mitsunobu [DIAD, triphenylphosphine or (4-dimethylaminophenyl)diphenylphosphine, tert-butyl glycolate] and in a low-temperature, sodium-hydride mediated alkylation (tert-butyl bromoacetate) to give the  $N^9$ -substituted derivative.

Peptide nucleic acids (PNA) are oligonucleotide analogues in which the sugar-phosphodiester backbone is replaced by a nucleobase-derivatized N-(2-aminoethyl)glycine chain. PNA oligomers have a number of properties (e.g. DNA and RNA recognizing ability) which make them very useful in antisense therapeutics and as diagnostics tools.

Until recently the assembly of PNA oligomers was based on a Merrifield solid-phase synthesis of Boc/Z-protected monomers<sup>1</sup>. Relatively few experience is available with other combination of protecting groups (e. g. Fmoc/Z<sup>2</sup>, monomethoxytrityl/anisoyl<sup>3</sup>) and there is a continuous need for deprotection conditions compatible with different protocols.

We would like to report herein on our results concerning the synthesis of PNA monomers (1-5, FIG) using Fmoc group for the backbone and various other groups for the heterobase N-protection in order to enable the oligomer synthesis to be performed on a peptide synthesizer using Fmoc strategy. Nucleobase-substituted acetic acids were

prepared as follows. Thymin-1-ylacetic acid was prepared according to a published procedure<sup>1</sup>. Cytosine was anisoylated and subsequently alkylated to give the corresponding acid. As this compound was very poorly soluble in most organic solvents the more lipophilic 4-tert-butylbenzoylated derivative was alkylated and the acid was obtained using the procedure by Will et al.<sup>3</sup>. The anisoylated adenine derivative was similarly prepared<sup>3</sup>. However, we have found more convenient to peranisoylate and then hydrolyze  $N^9$ -ethoxycarbonyladenine to afford the same acid.

For the guanine monomer  $N^2$ -isobutyryl- $O^6$ -diphenylcarbamoylguanine was chosen following the methodology by Zou and Robins<sup>4,5</sup> in order to avoid the formation of regioisomers. The persilylated derivative of the above compound can effectively be glycosylated or alkylated with alkoxymethyl halides<sup>5</sup> at  $N^9$  but it does not undergo reaction with *tert*-butyl bromoacetate neither in the absence nor in the presence of mercury(II) cyanide<sup>6,7</sup>. We then investigated the Mitsunobu reaction<sup>8</sup> of  $N^2$ -isobutyryl- $O^6$ -diphenylcarbamoylguanine with *tert*-butyl glycolate<sup>9</sup>. The resulting ester was obtained along with a negligible amount of the  $N^7$  regioisomer. It was difficult to remove completely the triphenylphosphine oxide byproduct therefore (4-dimethylaminophenyl)-diphenylphosphine<sup>10</sup> was used which gave an acid-removable phosphine oxide<sup>11</sup>. Alternatively, low-temperature, sodium hydride-mediated alkylation with *tert*-butyl bromoacetate gave the product with excellent  $N^9/N^7$  selectivity. When the alkylation was performed near room temperature significant amounts of  $N^7$  isomer was formed. The free acid was then obtained in an acidic hydrolysis of the *tert*-butyl group.

The coupling of the appropriate nucleobase-substituted acetic acids with the common backbone tert-butyl N-[2-(N-9-fluorenylmethoxycarbonyl)aminoethyl]glycinate hydrochloride<sup>2</sup> was performed using standard methods of peptide synthesis.

The synthesis of some new PNA oligomers from our monomers is under way.

#### REFERENCES

- 1. Dueholm, K. L.; Egholm, M.; Behrens, C.; Christensen, L.; Hansen, H. F.; Vulpius, T.; Petersen, K. H.; Berg, R. H.; Nielsen, P. E.; Buchardt, O., J. Org. Chem., 1994, 59, 5767-5773.
- Thomson, S. A.; Josey, J. A.; Cadilla, R.; Gaul, M. D.; Hassman, C. F.; Luzzio, M. J.; Pipe, A. J.; Reed, K. L.; Ricca, D. J.; Wiethe, R. W.; Noble, S. A., *Tetrahedron*, 1995, 51, 6179-6194.

- Will, D. W.; Langner, D.; Knolle, J.; Uhlmann, E., Tetrahedron, 1995, 51, 12069-12082.
- 4. Zou, R.; Robins, M. J., Can. J. Chem., 1987, 65, 1436-1437.
- Robins, M. J.; Zou, R. M.; Guo, Z. Q.; Wnuk, S. F., J. Org. Chem., 1996, 61, 9207-9212.
- 6. Kim, C. U.; Misco, P. F.; Luh, B. Y.; Martin, J. C., Tetrahedron Lett., 1990, 31, 3257-3260.
- Kim, C. U.; Misco, P. F.; Luh, B. Y.; Hitchcock, M. J. M.; Ghazzouli, I.; Martin, J. C., J. Med. Chem., 1991, 34, 2286-2294.
- 8. Jenny, T. F.; Schneider, K. C.; Benner, S. A., Nucleosides Nucleotides, 1992, 11, 1257-1261.
- 9. Kricheldorf, H. R.; Kaschig, J., Liebigs Ann. Chem., 1976, 882-890.
- Brune, H. A.; Falck, M.; Hemmer, R.; Schmidtberg, G.; Alt, H. G., Chem. Ber., 1984, 117, 2791-2802.
- 11. von Itzstein, M.; Mocerino, M., Synth. Commun., 1990, 20, 2049-2057.